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Key indicators

Single-crystal X-ray study
 $T = 173\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$
 R factor = 0.032
 wR factor = 0.091
Data-to-parameter ratio = 10.4For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

2-Phenylisophosphindoline-1,3-dione

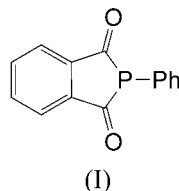
The title compound, $\text{C}_{14}\text{H}_9\text{O}_2\text{P}$, displays a near-planar isophosphindoline-1,3-dione skeleton with slightly displaced P and O atoms. Weak $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonding in the crystal structure results in a three-dimensional network.

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Comment

The title compound, (I), is a rare example of a crystallographically characterized dicarbonylphosphine. The only related compound whose crystal structure has been reported is bis(2,2-dimethylpropionyl)phenylphosphine, [$\text{BuC}(\text{O})_2$] $\text{P}-\text{Ph}$, (II) (Becker *et al.*, 1985).

The isophosphindoline-1,3-dione skeleton in (I) is almost planar, in contrast to compound (II), where rotation around the $\text{P}-\text{C}$ bonds results in an almost orthogonal orientation of the two carbonyl groups with an $\text{O}-\text{C}\cdots\text{C}-\text{O}$ angle of 73.49° [no standard uncertainty available (Becker *et al.*, 1985)]. In compound (I), the P and O atoms deviate by -0.388 (2), 0.209 (3) and 0.204 (2) \AA from the $\text{C}1/\text{C}3/\text{C}7\text{A}/\text{C}8$ plane, most likely because of the steric demand of the phenyl group. In the absence of the phenyl group, as is the case in the isophosphindolyl-1,3-dione anion (Liotta *et al.*, 1984), planarity of the same molecular fragment is observed. The geometry at the P atom is trigonal-pyramidal in compound (I), indicating a stereochemically active electron pair. This is in contrast to the nitrogen analogue of (I) (Magomedova *et al.*, 1981), where the N atom is in a trigonal-planar environment. In the crystal lattice, 2-phenylisophosphindoline-1,3-dione

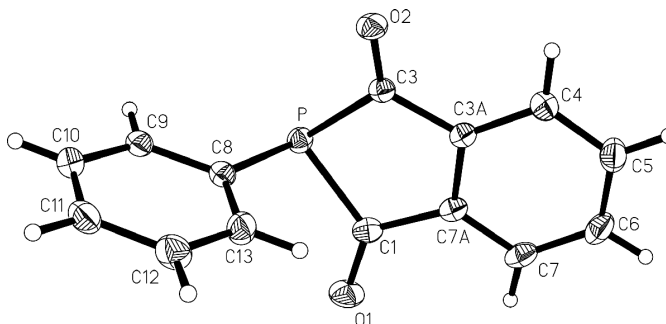


Figure 1

A view of the title compound, with displacement ellipsoids drawn at the 30% probability level.

employs both carbonyl O atoms for C—H···O hydrogen bonding (see Table 1), resulting in a three-dimensional network.

Experimental

The title compound was prepared according to a published procedure (Fenske *et al.*, 1976).

Crystal data

C₁₄H₉O₂P
M_r = 240.18
 Monoclinic, *P*2₁/*n*
a = 9.4273 (15) Å
b = 14.479 (2) Å
c = 9.4762 (15) Å
 β = 117.662 (2)°
V = 1145.7 (3) Å³
Z = 4

D_x = 1.393 Mg m⁻³
 Mo *K*α radiation
 Cell parameters from 3469 reflections
 θ = 2.5–25.9°
 μ = 0.22 mm⁻¹
T = 173 (2) K
 Plate, yellow
 0.40 × 0.30 × 0.05 mm

Data collection

Bruker SMART 1000/P4
 diffractometer
 ω and φ scans
 Absorption correction: none
 5722 measured reflections
 1970 independent reflections

1736 reflections with *I* > 2σ(*I*)
*R*_{int} = 0.021
 θ_{\max} = 25.0°
h = -11 → 10
k = -16 → 17
l = -10 → 11

Refinement

Refinement on *F*²
R[*F*² > 2σ(*F*²)] = 0.032
wR(*F*²) = 0.091
S = 1.07
 1970 reflections
 190 parameters
 All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0543P)^2 + 0.297P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.37 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\min} = -0.16 \text{ e } \text{Å}^{-3}$

Table 1

Hydrogen-bonding geometry (Å, °).

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C4—H4···O2 ⁱ	0.96 (2)	2.59 (2)	3.517 (2)	163.5 (14)
C7—H7···O2 ⁱⁱ	0.91 (2)	2.37 (2)	3.201 (2)	150.5 (17)
C10—H10···O1 ⁱⁱⁱ	0.90 (2)	2.53 (2)	3.368 (2)	155.5 (16)

Symmetry codes: (i) 1 - *x*, -*y*, -*z*; (ii) $\frac{1}{2} - x, \frac{1}{2} + y, \frac{1}{2} - z$; (iii) $\frac{1}{2} - x, y - \frac{1}{2}, \frac{1}{2} - z$.

H atoms were found in difference Fourier maps and refined with isotropic displacement parameters. The final C—H distances are in the range 0.91 (2)–0.96 (2) Å.

Data collection: *SMART* (Bruker, 1997–1999); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1997–1999); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.

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